Interactions Between SO₃, HCI, HBr, PM and Trona Injection in DSI

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Regulations

MATS - coal-fired boilers

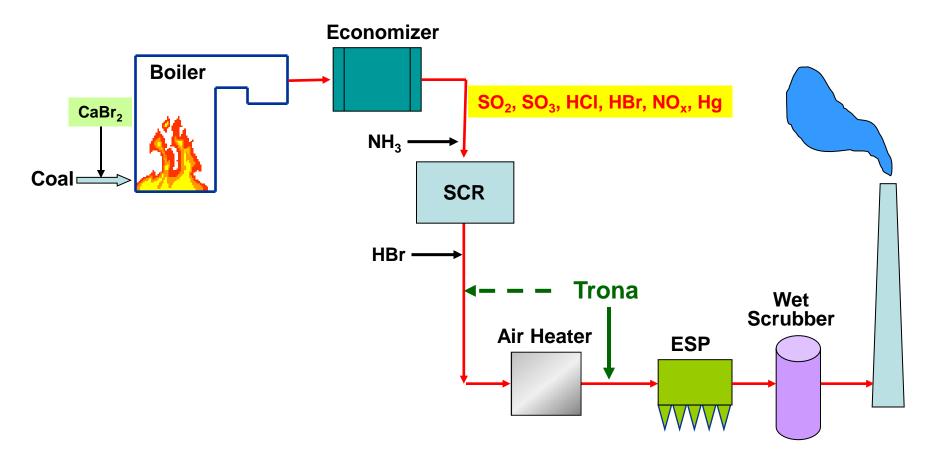
- HCI: 0.002 lb/mmBTU
- Mercury: 1.2 lb/TBTU
- PM: 0.03 lb/mmBTU

Proposed Industrial Boiler MACT – solid fuel

•HCI: 0.022 lb/mmBTU



Trona Injection to Mitigate Acid Gases





Trona Calcination

 $2(Na_2CO_3 \cdot NaHCO_3 \cdot 2H_2O) \text{ (s) } \rightarrow 3Na_2CO_3(s) + 5H_2O(g) + CO_2(g)$

◆ Acid Gas Neutralization Reactions $Na_2CO_3 + SO_2 + 1/2O_2 \rightarrow Na_2SO_4 + CO_2$ $Na_2CO_3 + SO_3 \rightarrow Na_2SO_4 + CO_2$ $Na_2CO_3 + 2HCI \rightarrow 2NaCI + H_2O + CO_2$ $Na_2CO_3 + 2HBr \rightarrow 2NaBr + H_2O + CO_2$



Chemical Reactions in Flue Gas

$\diamond \operatorname{Na_2CO_3} + \operatorname{SO_3} \rightarrow \operatorname{Na_2SO_4} + \operatorname{CO_2}$

• Sometimes SO_3 is used to condition fly ashes to enhance ESP performance.

 Trona is effective in mitigating SO₃ and can lower the ash resistivity thus conditioning the ESP. However trona does add the dust load for ESP and may cause a problem on undersized units.

$\diamond \text{Na}_2\text{CO}_3 + 2\text{HCI} \rightarrow 2\text{NaCI} + \text{H}_2\text{O} + \text{CO}_2$

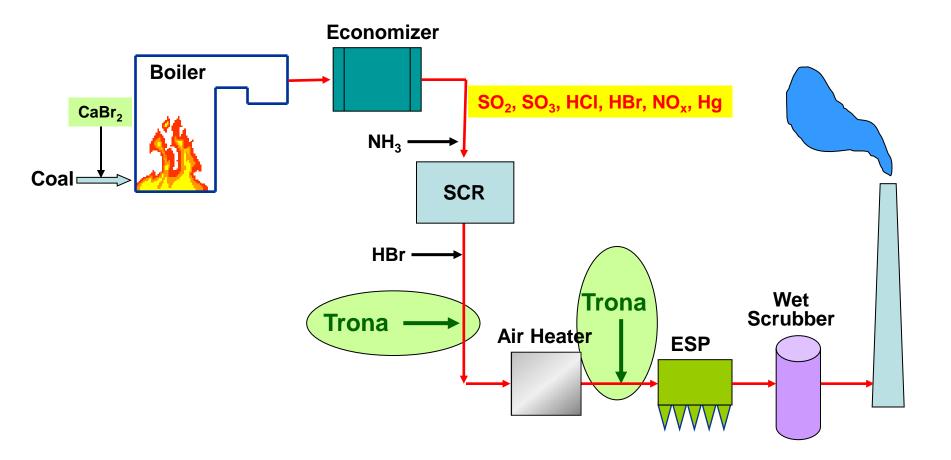
 $\diamond \text{Na}_2\text{CO}_3 + 2\text{HBr} \rightarrow 2\text{NaBr} + \text{H}_2\text{O} + \text{CO}_2$

• HCI and HBr can oxidize mercury, thus enhance mercury removal.

• HCl is the surrogate for all acid gases in both MATS (Utility MACT) and Boiler MACT.



Solution: Location!





Summary

Flue gas treatment has become a chemical processing plant

- Treating one component can affect others
- Good understanding of chemistries helps
- Apply system approach

All parties need to work closely, especially the one who designs and integrates the system should involve all suppliers in the trial and design of permanent systems.



Questions?

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