



Graymont Marketing Group

# Understanding Hydrated Lime Properties In Acid Gas Control

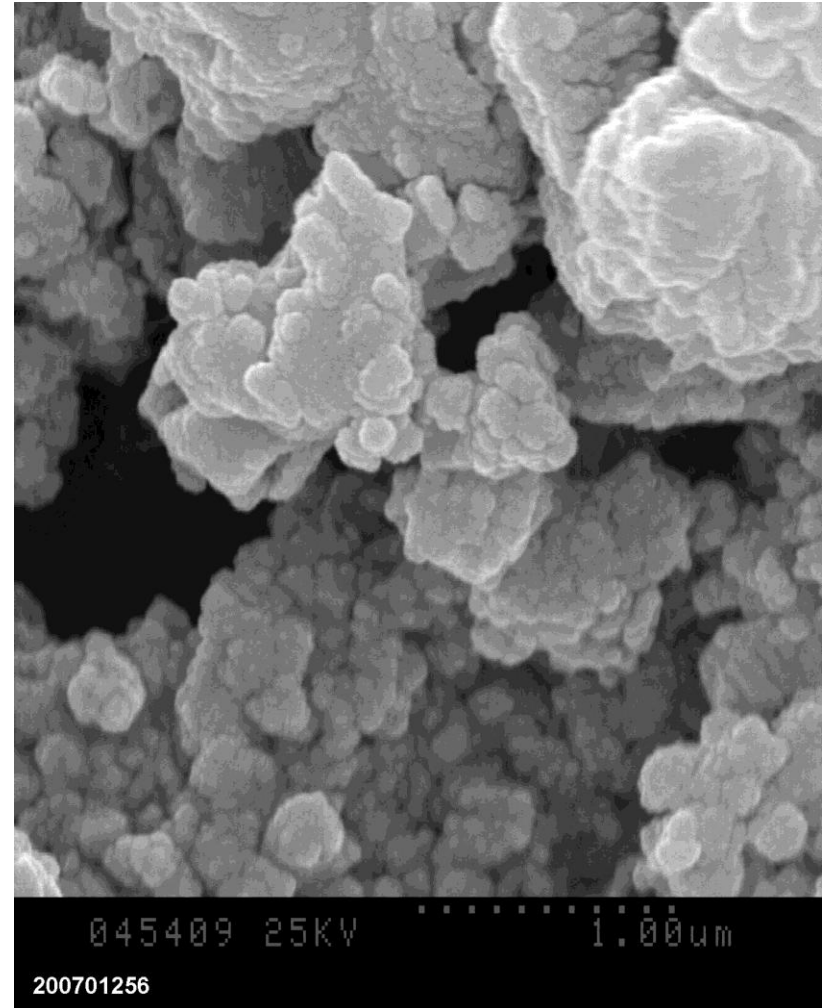
Mike Tate

McIlvaine Company Hot Topic Hour

June 21, 2012

# Understanding Sorbent Properties

- Material Handling
- Chemistry
- Reactivity
- Effect on Particulate Removal
- Reaction Products



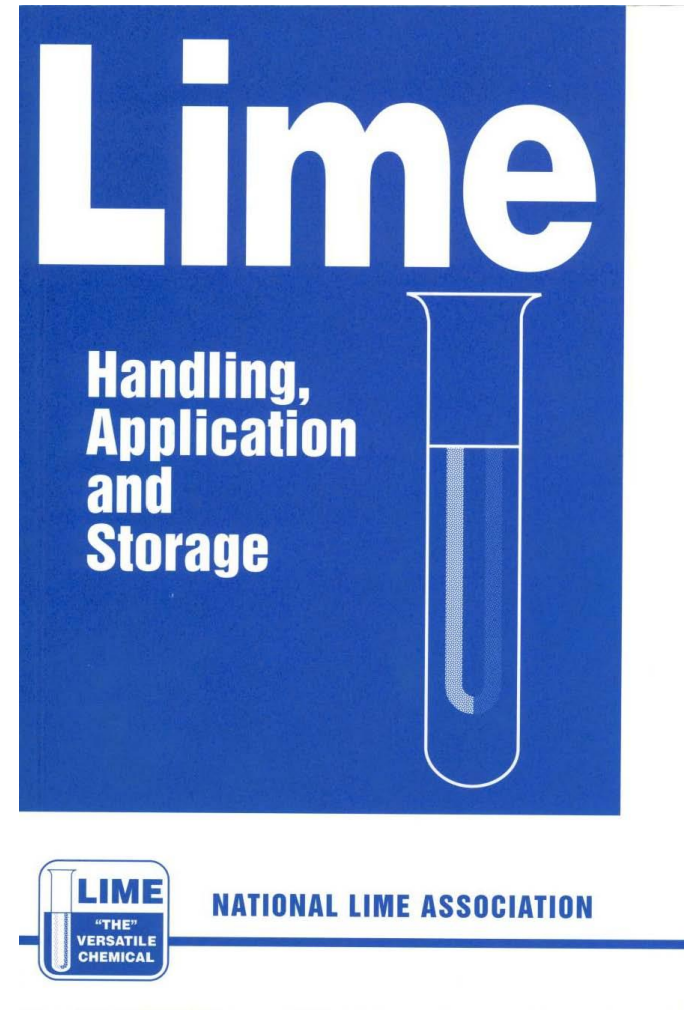


**GRAYMONT**

# Material Handling

# Material Handling Characteristics

- **Silo Design**
  - Density (25 – 35 lbs/ft<sup>3</sup>)
  - Cone Angle
  - Flowability
    - Flow Aids
    - Rotary Feeder
- **Pneumatic Conveying**
  - Air Temperature Effect
  - Moisture Level
  - Piping design





**GRAYMONT**

# Chemistry

# Chemical Composition

- Mining location can influence the chemistry of the sorbent.
- Levels of calcium and magnesium compounds are important. (90% min.)
- Higher purity – Better?





**GRAYMONT**

# Reactivity

# Sorbent Reactivity

- Normalized Stoichiometric Ratio
- Fineness/Surface Area
- Temperature
- Residence Time





# Sorbent Molecular Weight Analysis

Sorbent	Chemical Formula	Moles to Capture 1 Mole SO <sub>2</sub>	Molecular Weight
Hydrated Lime	Ca(OH) <sub>2</sub>	1	74
Trona	Na <sub>2</sub> CO <sub>3</sub> • NaHCO <sub>3</sub> • 2H <sub>2</sub> O	.67	151
Sodium Bicarbonate	NaHCO <sub>3</sub>	2	168

## Example: Trona Reactions

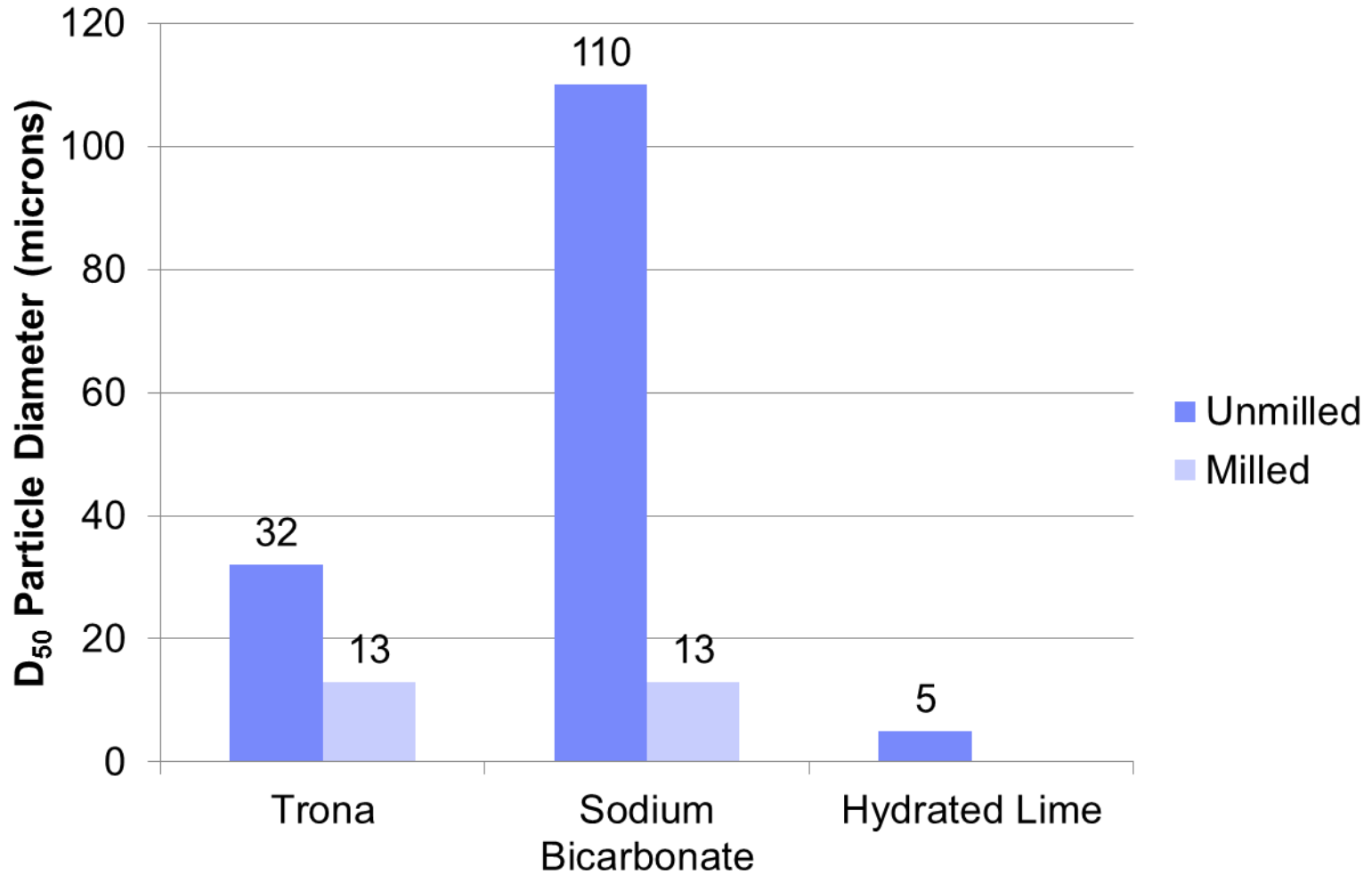
**Primary Reaction** -  $2(\text{Na}_2\text{CO}_3 \cdot \text{NaHCO}_3 \cdot 2\text{H}_2\text{O})(\text{s}) + \text{heat} \rightarrow 3\text{Na}_2\text{CO}_3(\text{s}) + 5\text{H}_2\text{O}(\text{g}) + \text{CO}_2(\text{g})$

**Secondary Reaction** -  $\text{Na}_2\text{CO}_3 + \text{SO}_2 \rightarrow \text{Na}_2\text{SO}_3 + \text{CO}_2$

# NSR – Ratio of Moles Not Weight

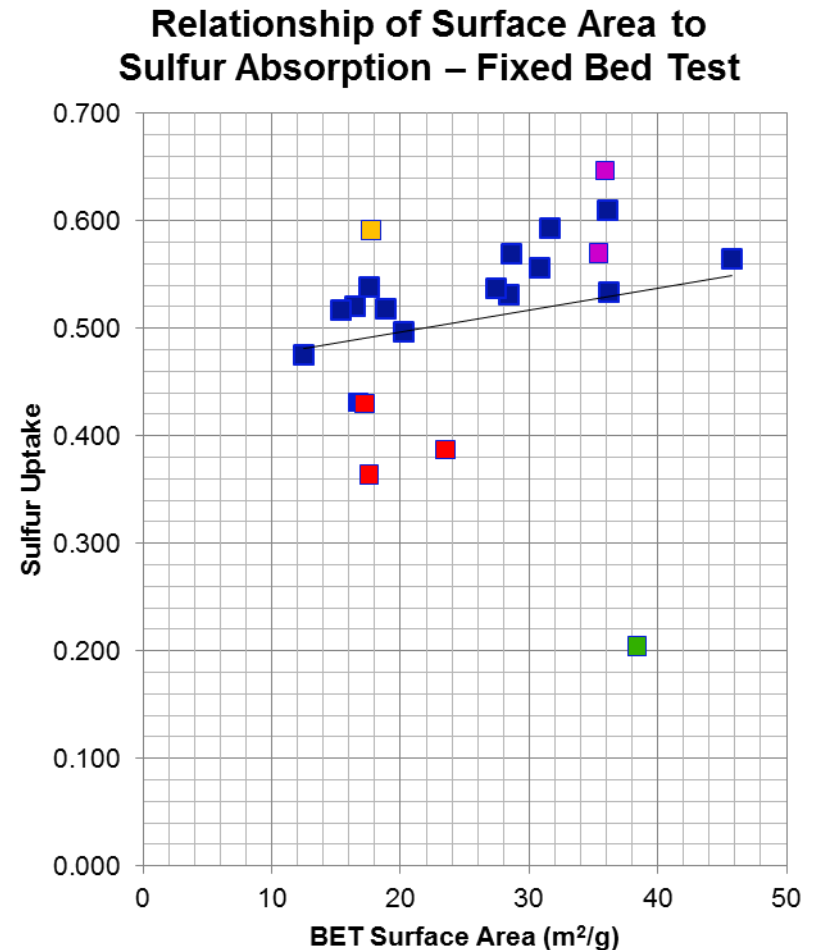
Normalized Stoichiometric Ratio	Theoretical Ratio By Weight of Sorbent to SO <sub>2</sub> for Complete Capture		
	Hydrated Lime	Trona	Sodium Bicarbonate
1.0	1.16	2.35	2.63
1.5	1.73	3.53	3.94
2.0	2.31	4.71	5.25

# Sorbent Particle Size



# Sorbent Surface Area

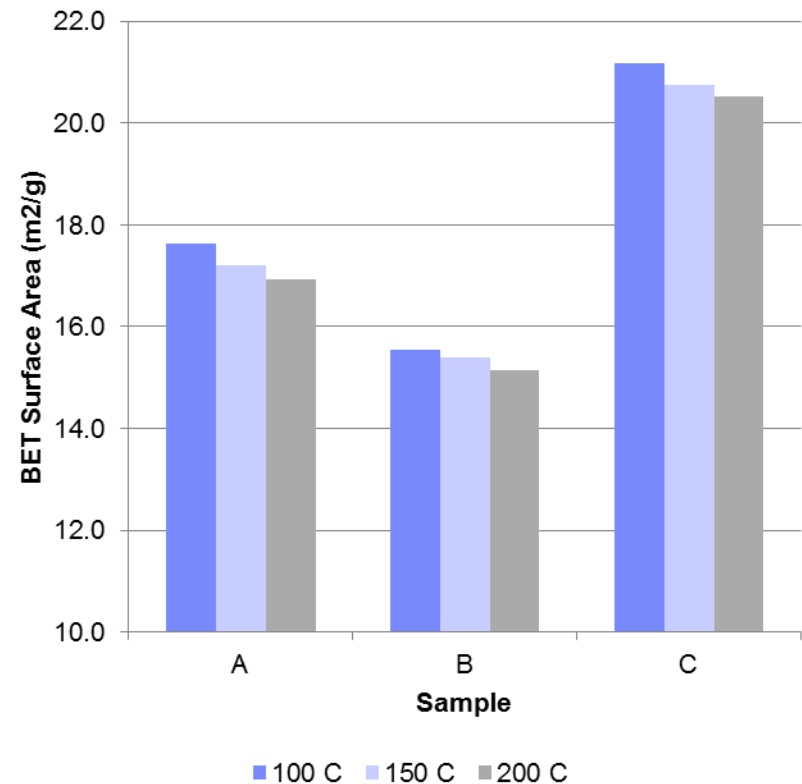
- As surface area increases, sulfur removal increases?
- A dynamic variable
  - Temperature Effects
  - Pore plugging



# Surface Area Measurement

- No standards currently exist for sample preparation.
- Begun working on standards for sample preparation in ASTM
- Temperature and length of time for sample preparation can influence results
- Vacuum preparation may also be a factor

**Effect of Sample Drying Temperature on Surface Area Measurement**



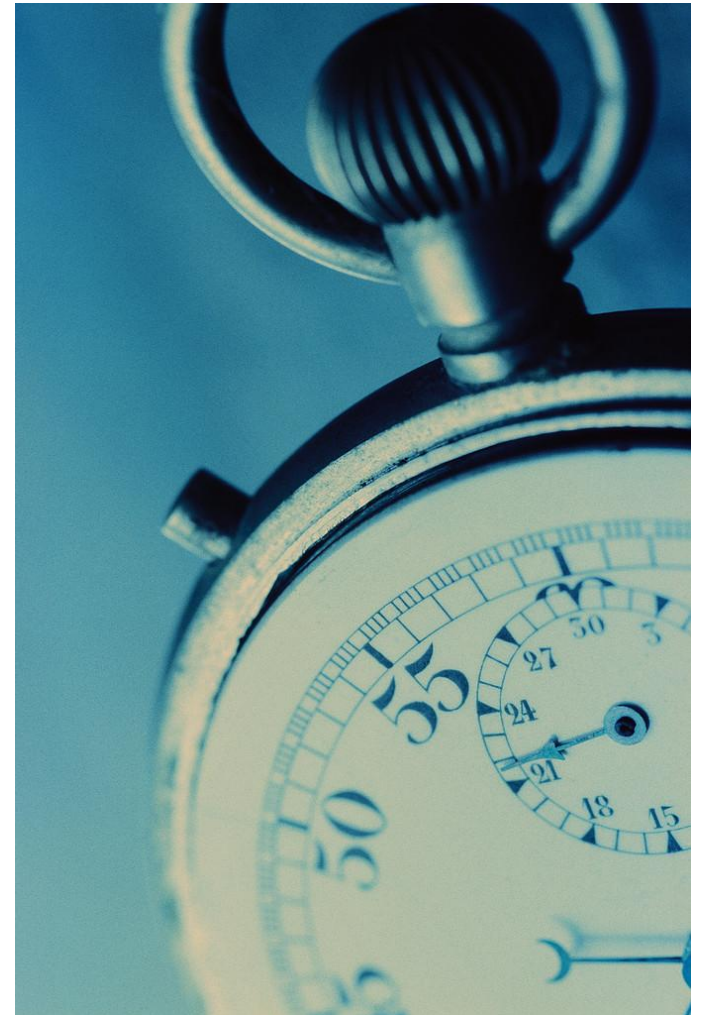
# Temperature Effects

---

- Flue gas contains multiple components that could potentially react with the sorbent.
- These components react with the sorbent in different temperature windows.
- Downstream effects of the sorbent and the reaction products must be considered.
- Trade-offs to get the targeted level of removals must be evaluated on a site by site basis.

# Residence Time Impact

- Longer times of exposure benefit all sorbents.
- Temperature profile of exposure time is important.





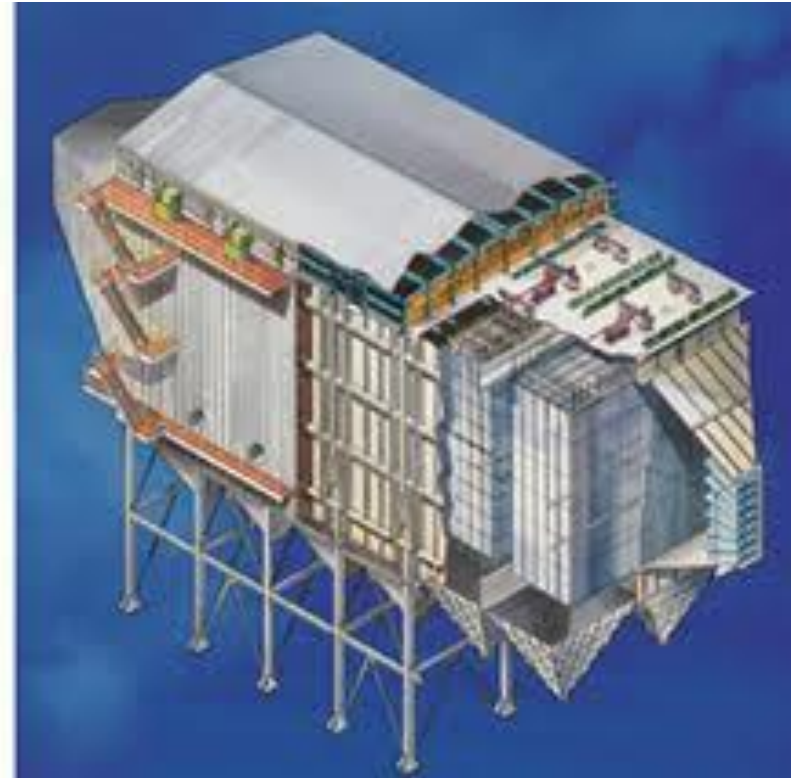
**GRAYMONT**

# Particulate Removal



# Particulate Removal Considerations

- **Baghouse**
  - Provides greater exposure of the sorbent to flue gas which can improve removal efficiencies
- **Electrostatic Precipitator**
  - Increased solids concentration can be problematic for ESP performance
  - Lime injection can increase resistivity. Humidification can improve collection performance.





**GRAYMONT**

# Reaction Products

## Effect On Fly Ash

- Spent sorbent can be a significant portion of the fly ash collected
- Soluble salts can limit beneficial use
- Solubility of metals in ash could be impacted





**GRAYMONT**

# Summary

# Summary

---

- There are a wide range of characteristics that can influence the performance of sorbents
- Differences in flue gas composition and process characteristics between boilers point to the need for testing to define performance
- The impact of reaction products should be considered when looking at the viability of different sorbent options



**GRAYMONT**

Questions?