

# Process gas cleanup from Coal and Municipal Gasification

By

Duane S Abbott

Beltran Technologies, Inc.

April 28<sup>th</sup>, 2011

---

**Beltran Technologies, Inc.  
1133 East 35 Street  
Brooklyn, N.Y. 11210**



- **Founded in 1946**
  - **Same corporate location for 50 years**
  - **Over 1000 installations world-wide**
  - **Own building in New York for Sales, Design, and R & D activities.**
  - **Own manufacturing facility for critical components in Brooklyn, NY.**
  - **Sales offices in EU and Asia.**
-

## Markets – Applications & Locations

- Acid Mist Collection – Titanium Dioxide
- Non Ferrous Smelters – Copper, Zinc, Nickel, Molybdenum, Zirconium, Gold
- Spent Acid Recovery
- Power Boilers
- Incinerators- Haz. Waste, Sewage Sludge
- Gasification
- Steel & Coke
- Automotive Spray Paint Finishing
- Chemicals & Pharmaceuticals
- Pulp & Paper
- North & South America, Europe, Asia and Africa



# Gasification



Fuel + Heat

Yields low BTU Syngas

Syngas then is used for Power generation

Internal combustion

Gas turbine

---

The fuels can be nearly anything that has fuel value.

Lignite, municipal waste, wood chips, saw dust, biomass.

Syngas is formed when there is insufficient oxygen content to fully burn the fuel. The composition varies but usually contains large amounts of carbon monoxide and hydrogen. The fuels frequently contain sulfur resulting in the formation of sulfur dioxide.

---

Sulfur dioxide is frequently scrubbed out with an alkaline stream forming a sulfate that is usually disposed of resulting in increased operation cost. Low concentration SO<sub>2</sub> gas can be absorbed in a circulating medium that can be stripped of the SO<sub>2</sub> effectively concentrating the SO<sub>2</sub> to be used or sold. The strong SO<sub>2</sub> can be sent to a WSA or a conventional sulfuric acid plant to be turned into sulfuric acid.

WSA or Wet Sulfuric Acid plant oxidizes the SO<sub>2</sub> to SO<sub>3</sub> above the acid dew point. As the SO<sub>3</sub> cools it combines with H<sub>2</sub>O in the gas. The resulting acid is then condensed and sold as product.

---

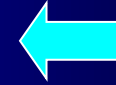
In a conventional acid plant the gas must be dried prior to oxidizing the  $\text{SO}_2$  to  $\text{SO}_3$ . The  $\text{SO}_3$  is then absorbed in a sulfuric acid tower.

Each process has its own set of benefits and must be evaluated on a case by case basis.

Particulate, mist, tars and soot formed in the gasification process can be removed effectively through wet electrostatic precipitation. Any acid or metal in the vapor phase will form submicron mist on cooling below the dew point of the material.

---

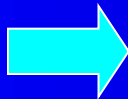
# BELTRAN TECHNOLOGIES, INC.



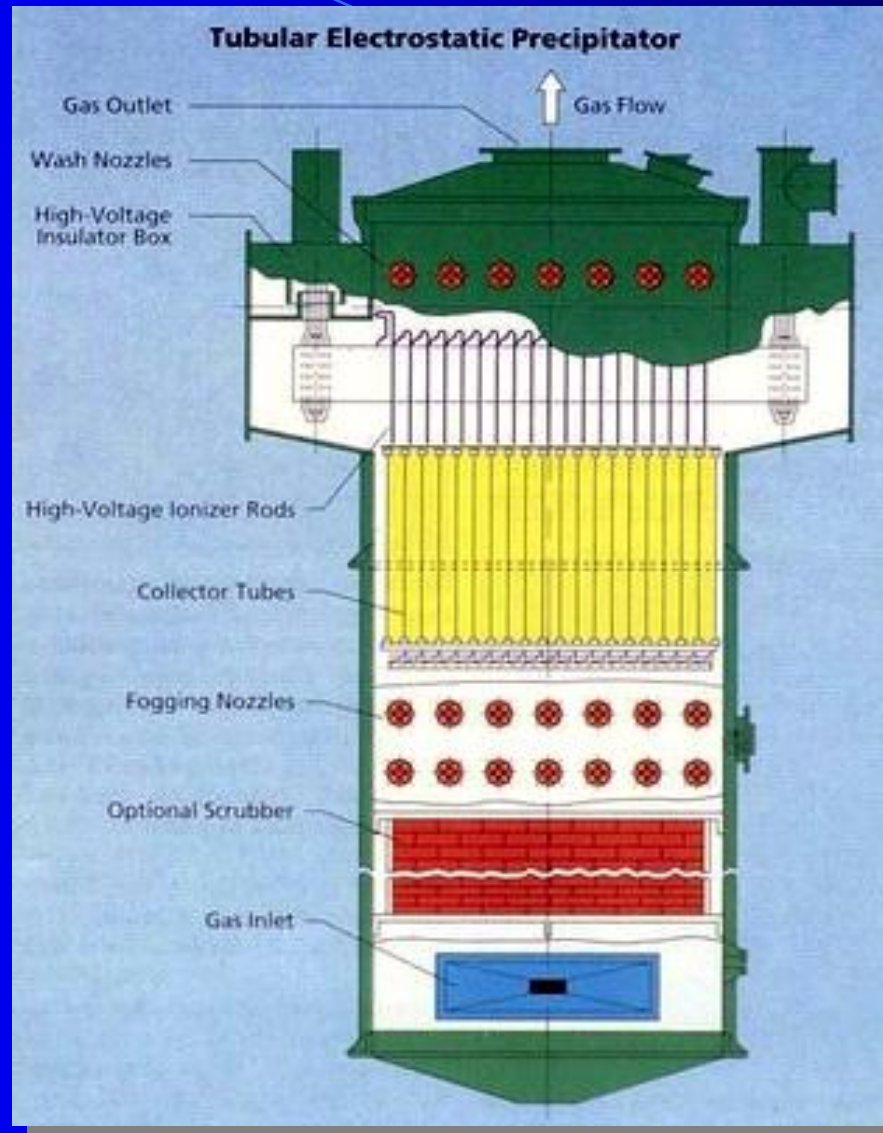
WESP OFF



WESP ON







# BELTRAN TECHNOLOGIES, INC.



# Wet Electrostatic Precipitator

## Design Comparison

- **CORROSION RESISTANT DESIGN** –  
Corrosion resistant alloy, FRP, and lead.
  - **SQUARE TUBE** vs. Round Tube -  
More Efficient Use of Space  
No wasted space, both inside and outside of tubes can be utilized with Beltran design
  - **SHORTER TUBE LENGTH** –  
Easier to maintain Ionizer alignment
  - **CONDUCTIVE FRP** over PP tube -  
Does not rely on water film for grounding, withstand temperature excursions better.
-

# Wet Electrostatic Precipitator

## Design Comparison

- **RIGID MAST ELECTRODES** -  
From alloy or lead.
  - **MECHANICALLY STRONGER** –  
Easy to clean, more corona discharge points per rod
  - **RIGID FRAME** –  
Does not elongate and move to affect ESP performance
  - **ISOLATED INSULATORS** –  
No stabilizing insulators at the bottom in gas stream
-

*BELTRAN TECHNOLOGIES, INC.*



# CONCLUSIONS

The application of the BELTRAN Wet Tubular Electrostatic Precipitator demonstrates that:

- ✓ Soot and mist formed from gasification can be removed with wet electrostatic precipitation.
  - ✓ WESPs used upstream of the fuel turbine or internal combustion engine can increase the life of the down stream equipment.
  - ✓ WESPs can effectively reduce smoke at the stack.
-