

McIlvaine Hot Topic Hour Evaporation and Crystallization of FGD Wastewater

September 3, 2009



Solutions & Technologies







Innovative Process Solutions Utilizing Evaporation & Crystallization as Core technologies



ZLD for FGD as practiced today

HPD

Brine

A2A S.p.A. (formerly Endesa Italia)

- Monfalcone Power Plant, Italy
- 336 MW Coal-Fired w/ LSFO Scrubber (MHI)
- **ZLD Operational Summer 2008**
- Dry Cake for Landfill Disposal

Lime-Soda Softener

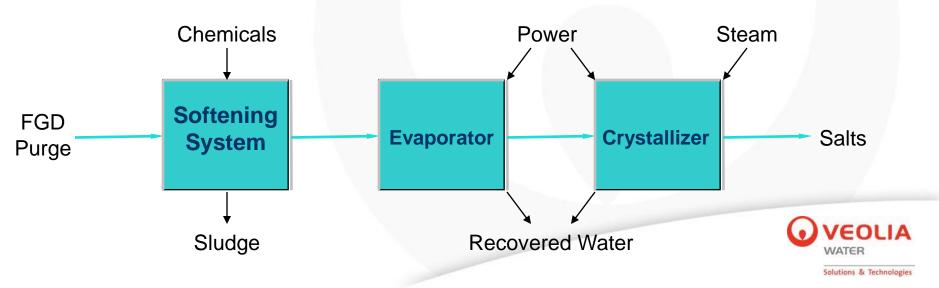




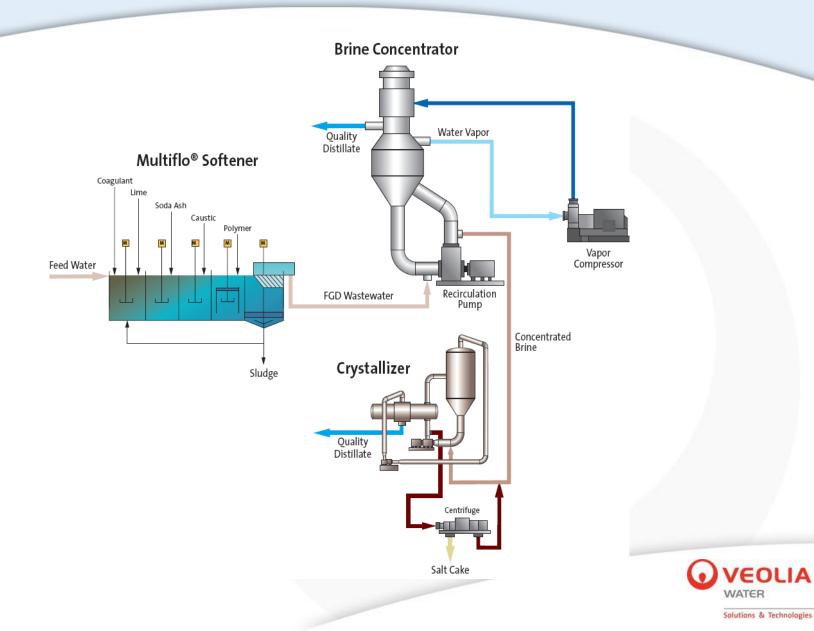
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Three Major Unit Operations

- A Softening System to reduce the overall solids and hardness content down to a level that is manageable for the evaporation system.
- A Falling Film Evaporator (MVR) to remove the major portion of water from the FGD purge stream.
- A Crystallizer (usually steam, can be MVR) to finish the process by removing the remainder of the water.



Requires chemicals, power & steam, produces sludge



Softening aids crystallization

Softening FGD wastewater with lime and soda ash substitutes sodium ions for most of the calcium and magnesium ions.

CaCl2 and MgCl2 have very high solubility – difficult to crystallize out. Softening changes the chemistry of FGD wastewater to NaCl, which crystallizes easily.





Softening Reactions



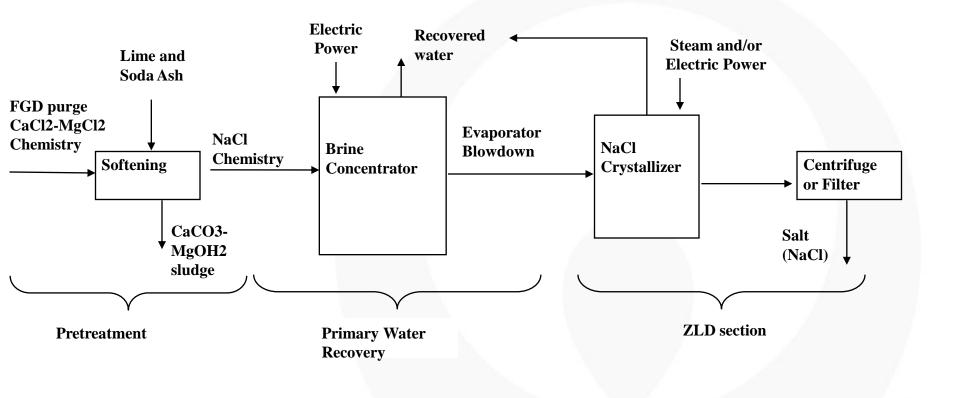
Mg⁺² + Ca(OH)₂ (lime)

 $Mg(OH)_2 + Ca^{+2}$ (sludge)

► $Ca^{+2} + Na_2CO_3 \longrightarrow CaCO_3 + 2 Na^{+1}$ (soda ash) (sludge)



Soften the FGD purge - create a NaCl based chemistry

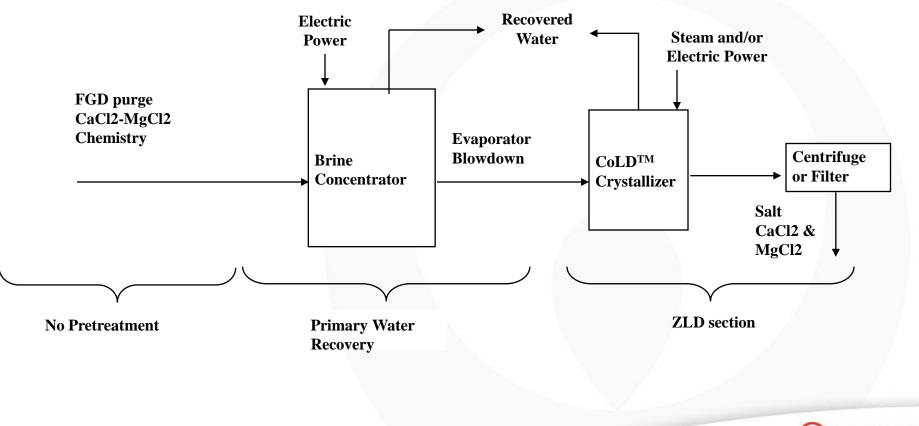




What's new?

HPD

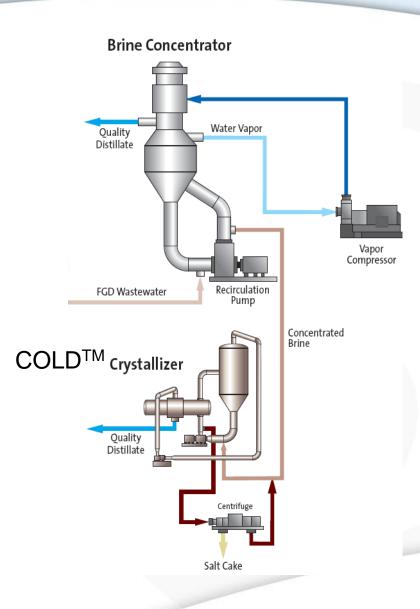
No pretreatment – Crystallizer based on industrial CaCl₂ production





Requires power and/or steam, no chemicals, reduced sludge





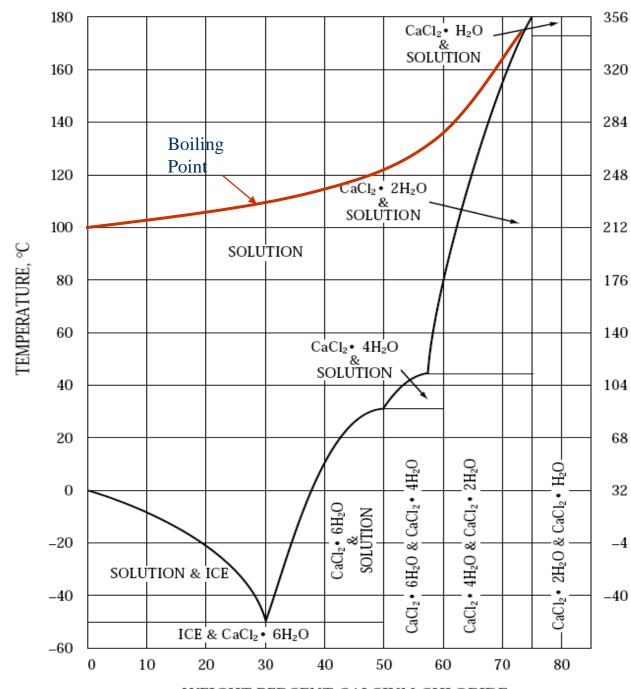


Two Options for Evaporation of FGD Wastewater

- Option 1: Pretreat FGD Wastewater Using Lime-Soda Ash softening
 - Replace Ca and Mg with Na
 - Chemistry is based on NaCl
- Option 2: Evaporate FGD Wastewater Directly
 - Chemistry is based on CaCl₂ and MgCl₂



At atmospheric pressure, the boiling point curve chases the solubility curve

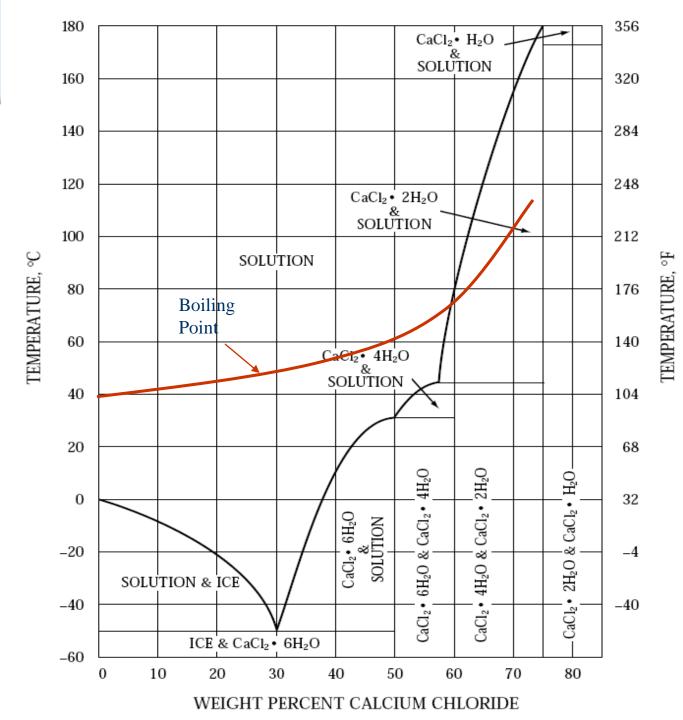


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FEMPERATURE,

WEIGHT PERCENT CALCIUM CHLORIDE

By lowering the pressure, the boiling point is lowered and a solid phase can form





Crystallization of high solubility salts at Low Temperature and Deep Vacuum



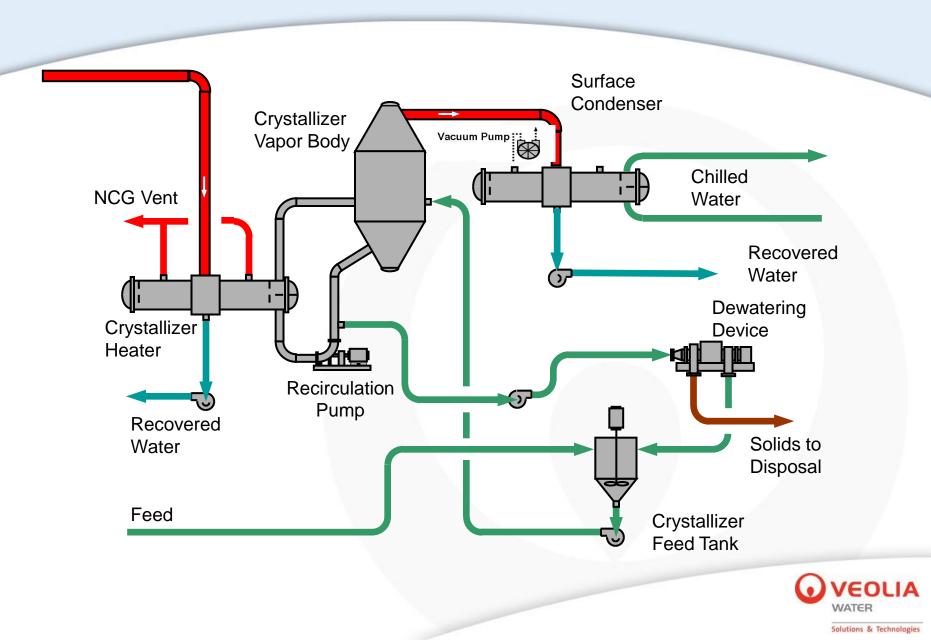
What is the CoLD[™] Process?

A conventional forced circulation crystallizer fitted with a vacuum system

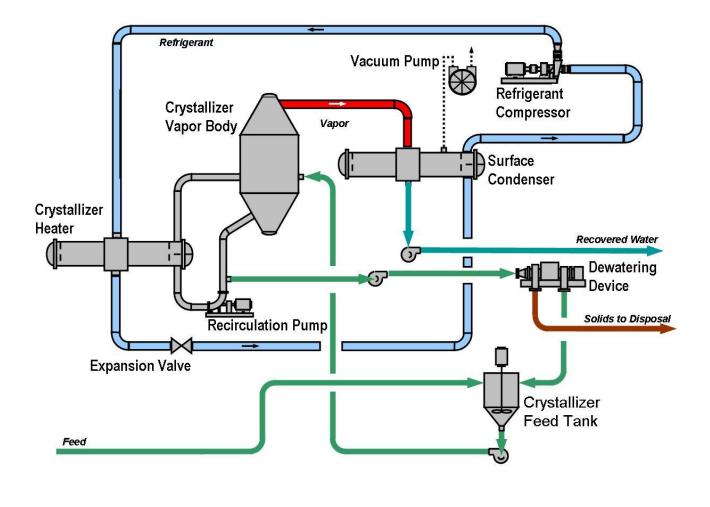
- Energy for evaporation can be provided by a conventional indirect heat pump, using a refrigerant fluid.
- Alternately, energy for evaporation can be provided by steam, and low temperature sink can be provided by chilled water.
- Low operating temperature lowers the solubility of the dissolved salts, so they crystallize at a lower concentration.
- CoLD[™] Process suited for waste streams containing highly soluble salts (e.g. FGD and IGCC blowdown, produced water, landfill leachate)



Steam Driven CoLD[™] Crystallizer



Indirect Heat Pump CoLD[™] Crystallizer HPD



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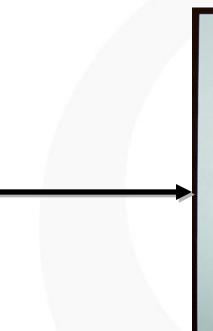
- No softening required => No chemicals required, no sludge produced.
- A stable, solid product is produced directly using only electric power.
- Chemical costs are eliminated and solids disposal costs are greatly reduced.
- Energy costs are still low.
- Complexity and footprint of equipment are greatly reduced.
- Materials of construction can be reduced



Result of CoLD[™] Process ZLD











Economic comparison of FGD ZLD Options

Facility Comparison

ZLD #1: 350 gpm capacity, Softening, Evaporator, Crystallizer

►ZLD #2: 350 gpm capacity, Evaporator, CoLDTM Crystallizer

	ZLD #1	ZLD #2
		_
Cap - Amor	\$5.4MM	\$5.4MM
0&M	\$2.8MM	\$2.8MM
Chemicals	\$6.3MM	\$250k
Disposal	\$4.5MM	\$1.0MM
Energy	\$1.8MM	\$2.3MM
Total Opex	\$15.4MM	\$6.4MM
Net Annual Cost	\$20.8MM	\$11.8MM
\$/gal	\$0.125	\$0.071
	NPV savings of \$88MM over 15 yrs at 7% discount rate	•





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