

*Natural Gas from Shale: the  
Resource of Our Time*

*Focus on the analysis of gasses in  
Water*

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- Introduction
  - “fracking” info
  
- Analyzing Methane and other gases
  - Headspace Theory and Operation
  - Method parameters and results

# The opportunity that lies ahead. . .

▶ “The United States is in the midst of a 21<sup>st</sup> century gold rush—a natural gas boom, driven in large part by the discovery of vast amounts of gas in unconventional plays. The spotlight is on the practice of fracking, which is quickly becoming one of the most highly visible and controversial environmental issues of the day...”\*

\* FrackingInsider.com, Kelley Drye & Warren LLP, last read August 10, 2011

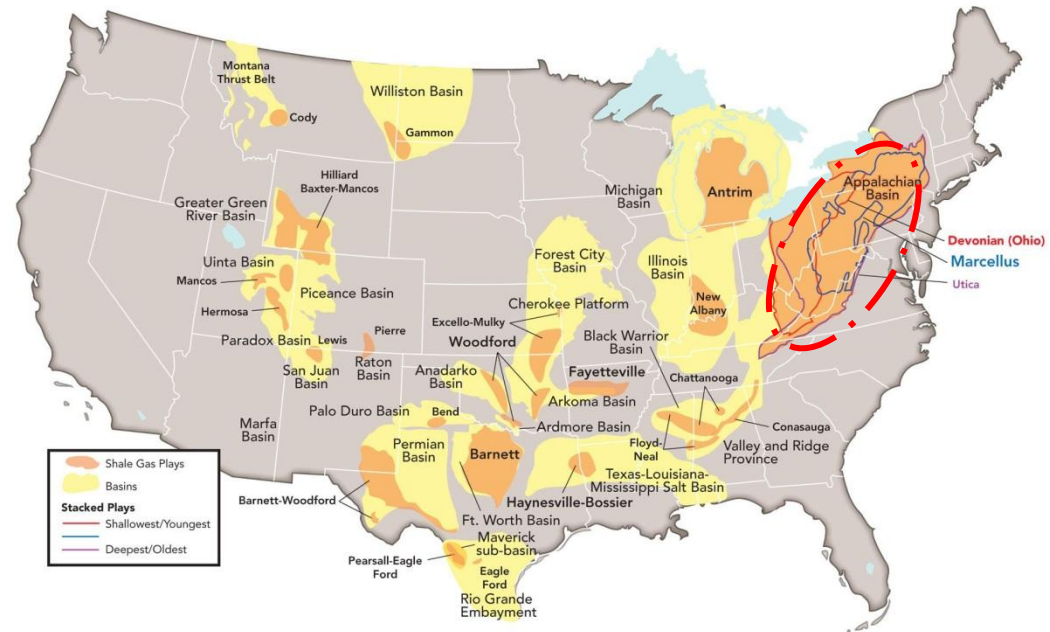
PerkinElmer can play a significant role in facilitating the environmental efficacy of the process – we have the know-how; the neutrality; the measurement tools to make people effective - making the process and results safe and profitable.



- ▶ Gas in shale lies deep in the earth
  - Typical wells are 5000-10000 ft deep
  - Shale is impervious, the gas is tightly held
- ▶ Horizontally drilled wells are required
  - Well is vertical until it nears the “pay zone”
  - Drilling is turned to horizontal – well understood process - ~50 years old
  - Continues through the deposit up to 1 mile.

## Shale Gas Opportunities

Shale gas plays, lower 48 states



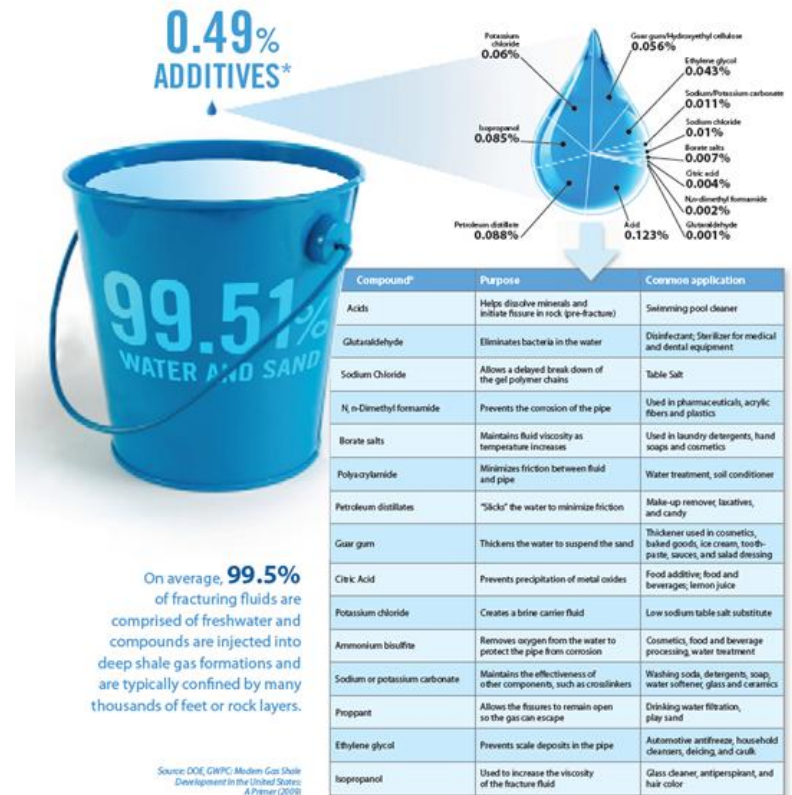
Source: Energy Information Administration based on data from various published studies. Updated March 2010.

## Testing is

- Inorganic
- Anions
- Methane in Water
- Radioactivity
- Little (yet) organic needs

Some of what goes into the well:

## A FLUID SITUATION: TYPICAL SOLUTION\* USED IN HYDRAULIC FRACTURING



On average, **99.5%** of fracturing fluids are comprised of freshwater and compounds are injected into deep shale gas formations and are typically confined by many thousands of feet or rock layers.

Source: DOE, GWPC: Modern Gas Shale Development in the United States: A Primer (2008)

\*The specific compounds used in a given fracturing operation will vary depending on source water quality and site, and specific characteristics of the target formation. The compounds listed above are representative of the major material components used in the hydraulic fracturing of natural gas shales. Compositions are approximate.

Additive volume = 25,000 gal in typical well

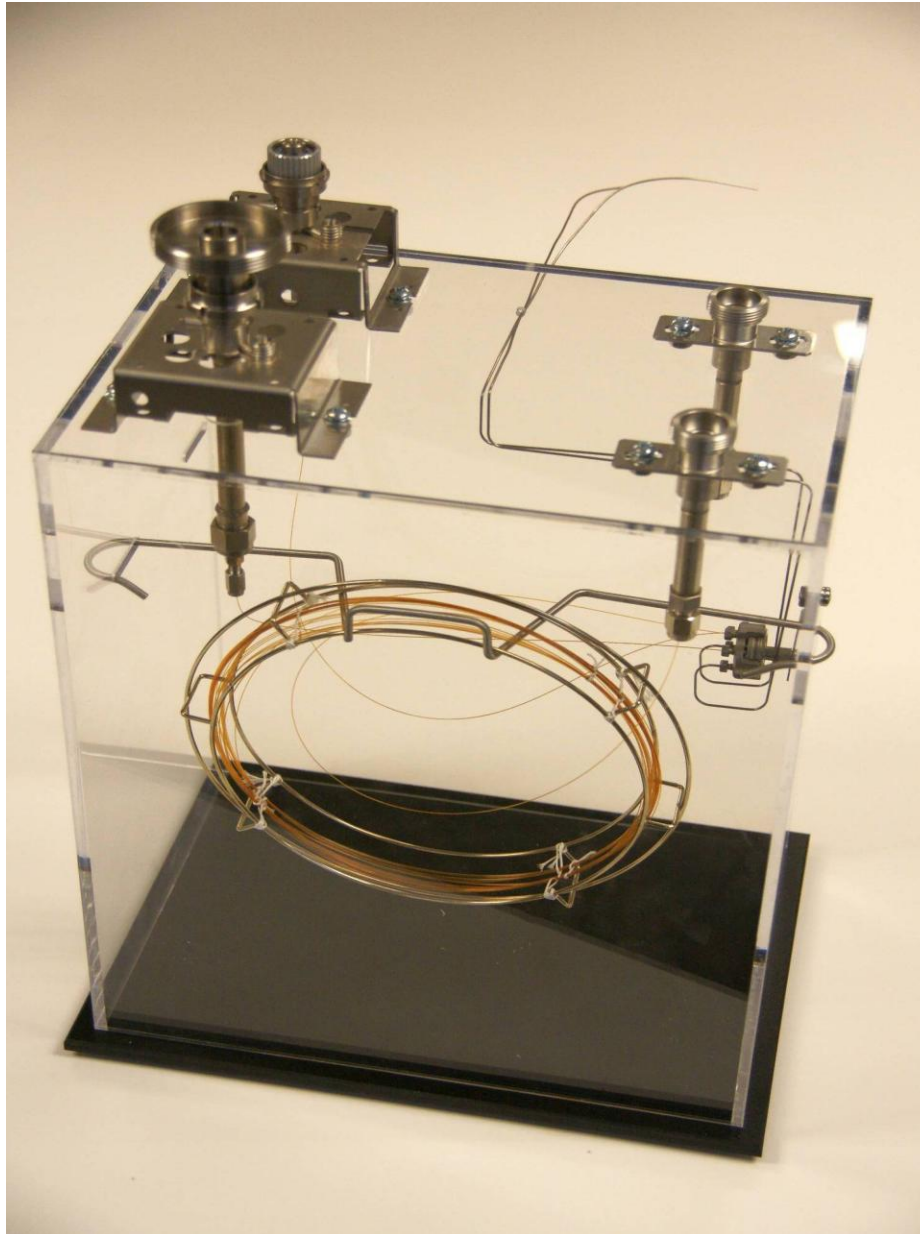
*The TurboMatrix Headspace  
and the Clarus 680 Gas Chromatograph*



**The Analysis of Methane and other  
gases in water using Headspace/  
Gas Chromatography  
Optimizing RSK 175**

*Theory and Operation*

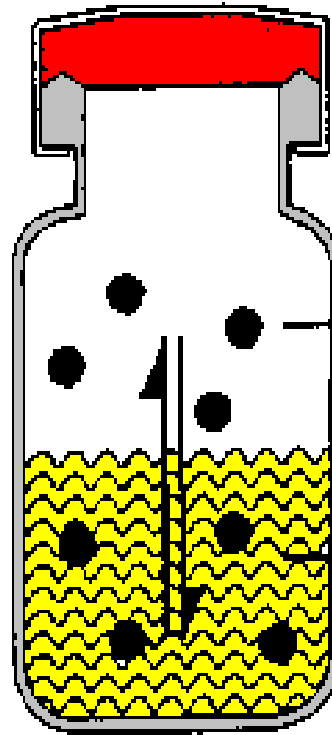




# Headspace: The sample Introduction System

Consider:

Time  
Temperature



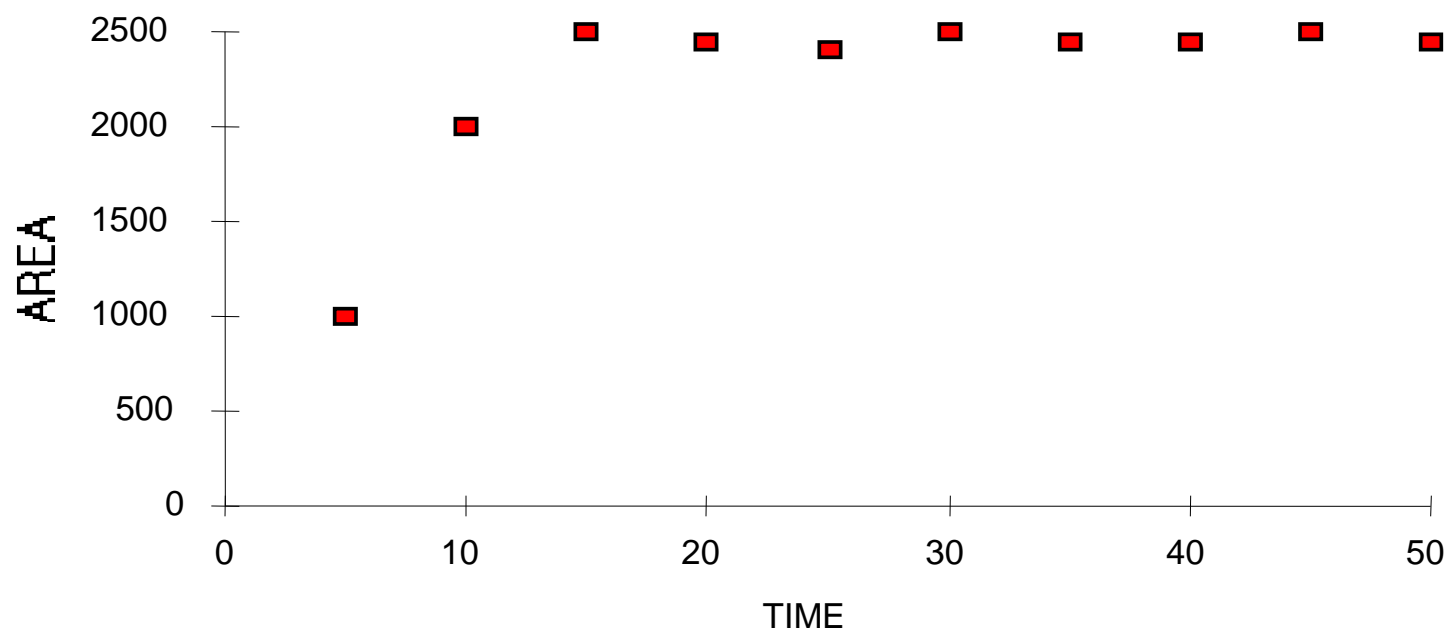
Solute "i"

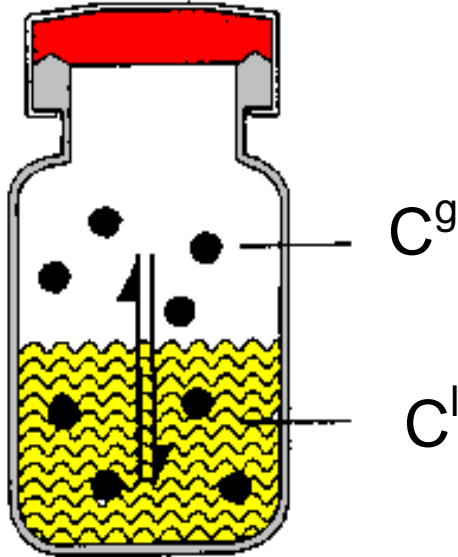
Liquid Sample



- ▶ Equilibration time must be long enough for each analyte to be in equilibrium
- ▶ Matrix effect – the partition coefficient is matrix dependent therefore the partition for each analyte in the standard and the sample must be the same

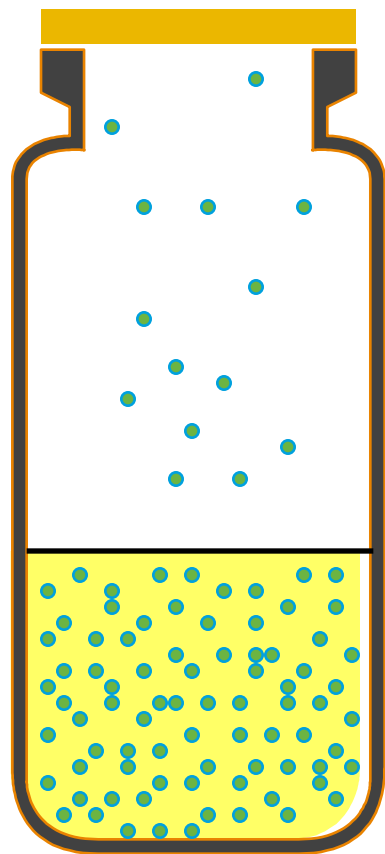
# Determining Equilibration Time



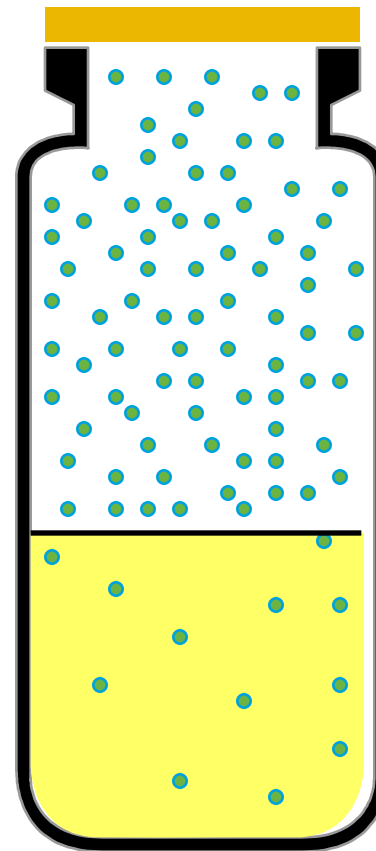


$$K = C^l / C^g$$

- K = Partition coefficient of a volatile
- $C^l$  = Concentration in the liquid phase
- $C^g$  = Concentration in the gas phase



When  $K$  is large

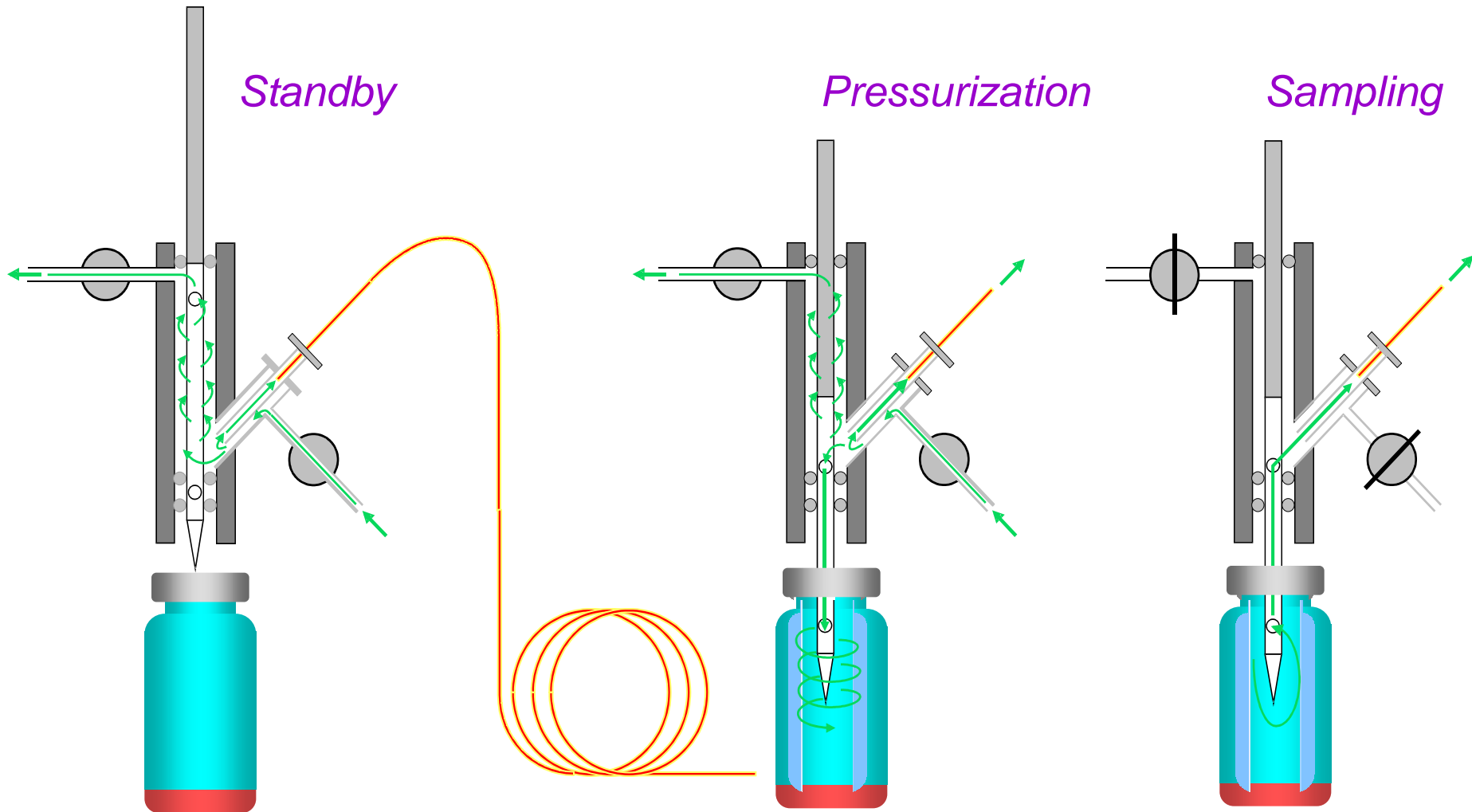


When  $K$  is small

$$A = \frac{C^0}{K + \beta}$$

- ▶ A = Peak Area or Response
- ▶  $C^0$  = Concentration of analyte in sample
- ▶ K = Partition Coefficient
- ▶  $\beta$  = Phase Ratio =  $V_g/V_s$
- ▶  $V_g$  = Volume of the gas phase
- ▶  $V_s$  = Sample Volume

# Balanced – Pressure Sampling

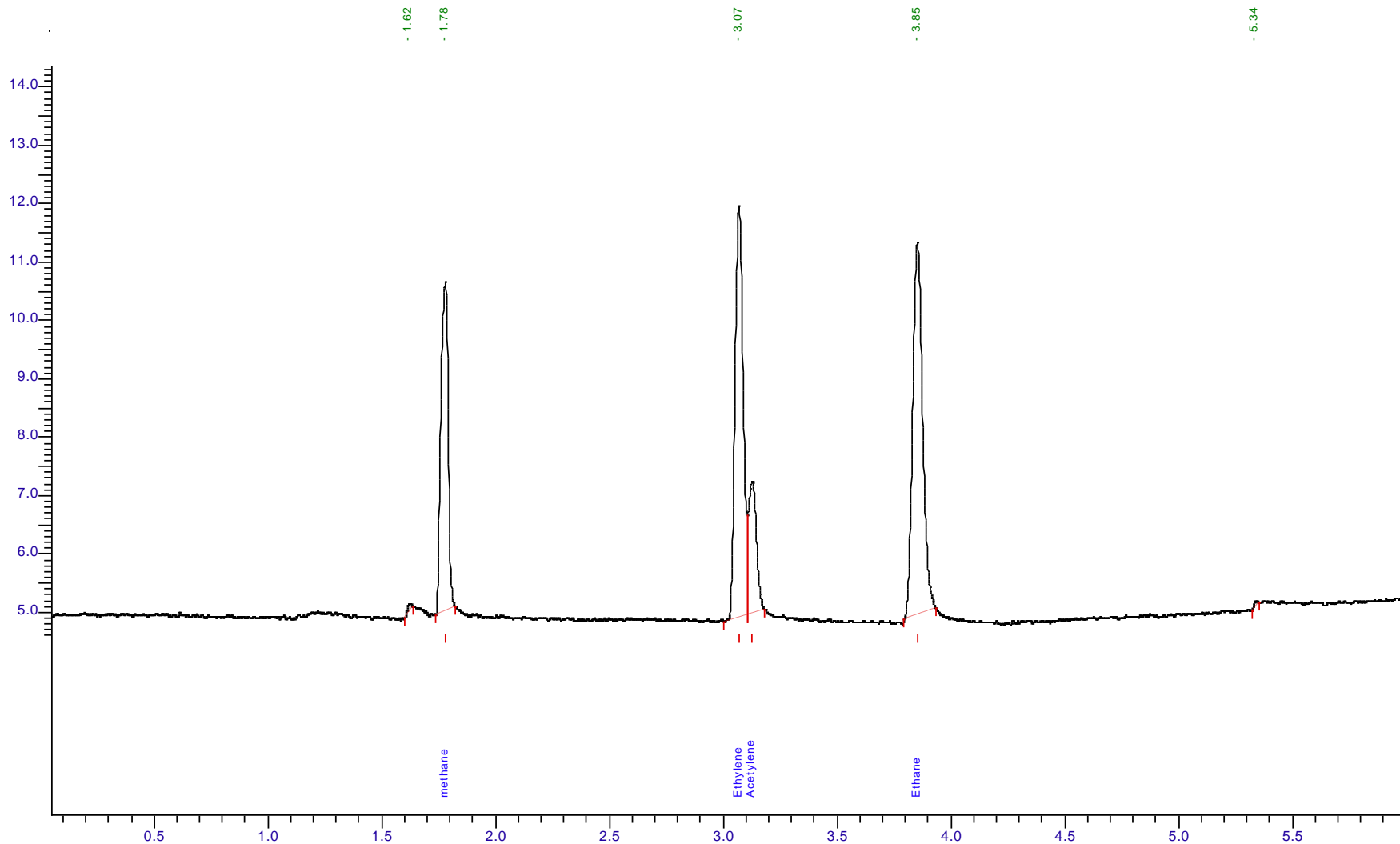






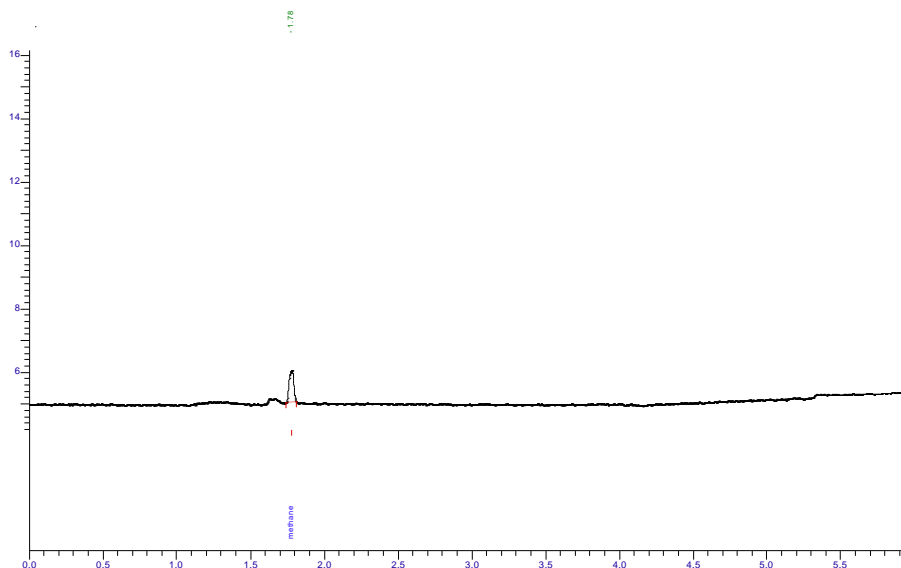
## *Method Parameters and Results*

# 10 ppb Chromatogram



Sample Name	Area (Methane)
15mL Water Blank	2093.5
15mL Water Blank	2163.7
15mL Water Blank	2337.4
15mL Water Blank	2124.3
Average	2179.7
%RSD	5%

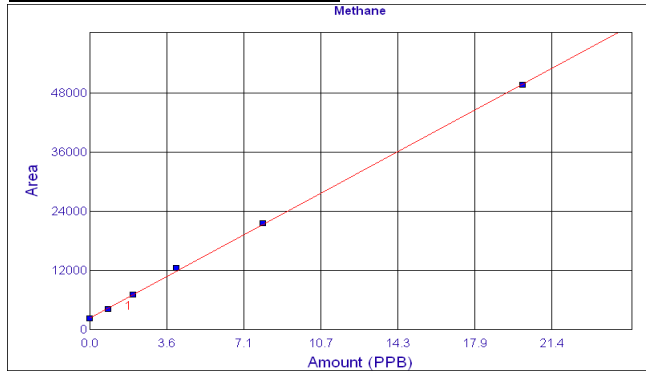
- ▶ Contribution is significantly below the reporting limit of 1ppb
- ▶ Incorporating this point on the curve as a “zero” amount incorporates a blank subtract



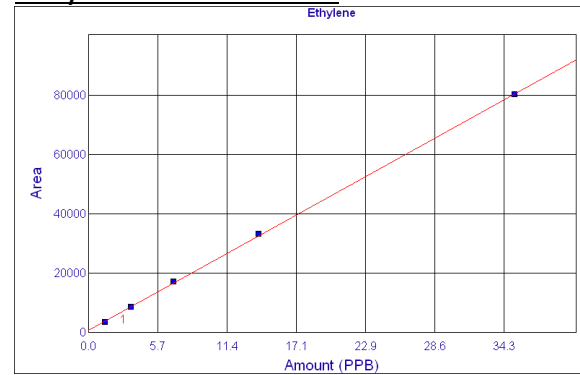
Level #	Methane	Ethylene	Ethane
Level 1	0.80	1.40	1.50
Level 2	2.00	3.50	3.75
Level 3	4.00	7.00	7.50
Level 4	8.00	14.00	15.00
Level 5	20.00	35.00	37.50

# Results of Calibration

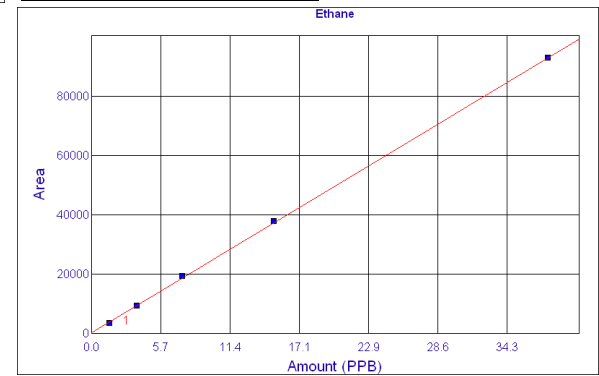
Methane:  $r^2 = 0.9996$



Ethylene:  $r^2 = 0.9998$



Ethane:  $r^2 = 0.9999$



# Quality Control Results

Methane		
Actual	Calc	% Dev
Amount	Amount	
2.00	2.05	2.50
10.00	10.72	7.20
14.00	15.19	8.50
20.00	20.69	3.45

Ethylene		
Actual	Calc	% Dev
Amount	Amount	
3.50	3.43	-2.00
17.50	18.68	6.74
24.50	26.40	7.76
35.00	36.44	4.11

Ethane		
Actual	Calc	% Dev
Amount	Amount	
3.75	3.59	-4.27
18.75	19.91	6.19
26.25	28.43	8.30
37.50	39.14	4.37



# Repeatability (n=5)

Concentration (ppb)	Methane Area	Ethylene Area	Ethane Area
8	43180	70067	80441
8	44330	70199	81390
8	43421	67911	79164
8	44331	71017	82016
8	42184	66722	76234
Average	43489	69183	79849
% RSD	2.1	2.6	2.9

- ▶ Fortunately, headspace/GC with Flame Ionization detection provides a very simple, fast, accurate and precise solution to this important analysis.

Thank you



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